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Subject Code:133502

GUJARAT TECHNOLOGICAL UNIVERSITY

BE- SEMESTER-III • EXAMINATION – SUMMER 2015

Date: 29/05/2015

Subject Name: Analytical Techniques Time: 02.30pm-05.00pm **Total Marks: 70 Instructions:** 1. Attempt all questions. 2. Make suitable assumptions wherever necessary. 3. Figures to the right indicate full marks. 07 **Q.1** Define following terms: Electromagnetic spectrum, Red shift, Finger print region, Hook's law, Degree of freedom, Nitrogen rule, Chromophore. What is chromatography? Draw clean diagram of HPLC and explain its **(b)** 02+02+03 applications. Write classification of chromatography and elaborate it with Thin layer **Q.2** 03+04chromatography. Specify the practices used for sampling of solid and liquid. **(b)** 03+0407 **(b)** Explain gravimetric analysis of iron. **Q.3** (a) Give details of precipitation, co-precipitation, post precipitation and precipitation 03+04from homogenous solution. Define errors, also state their classification and statistical representation of data. 02+02+03 **(b)** (a) Define following terms: normality, stoichiometry, titration, oxidation, molarity, Q.3 **07** saturated solution, strength Write detailed note on TGA (thermo-gravimetric analysis) and validation of **(b)** 03+04analytical methods. A sample of steel alloy is to be tested in the chemistry laboratory, which type of 02+05**Q.4** (a) titration do you think will be useful for its analysis, write the details of the method with chemical reactions. Write detailed notes on QC (quality control) and TQM (total quality **(b)** 03+04management). Explain principle of UV spectroscopy. Also explain its instrumentation and **Q.4** 03+04application. **(b)** What is retention factor, how it affects the separation also Elaborate factors 03+02+03 affecting R_f. Define mean, median, standard deviation, standard error, range, variance & 07 0.5 coefficient of variance. Write short note on: Mc Lafferty rearrangement, Principle of Mass Spectroscopy, 02+02+03 Fragmentation pattern OR Deduce the tentative structure of organic compound by using following data: 07 **Q.5** UV=transparent above 210 nm i) IR=3049-2624 cm⁻¹, 1445 cm⁻¹ ii) Molecular mass =88 iii) NMR (δ)= singlet at 8.122 (5.9 squares), triplet at 4.22 (12.2 square), iv) sextet at 1.77 (11.6 square), triplet at 1.33 (18.6 squares) **(b)** Explain good laboratory practices. **07**
