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## GUJARAT TECHNOLOGICAL UNIVERSITY

BE - SEMESTER-III(New) • EXAMINATION - WINTER 2016

Subject Code:2133605 Date:04/01/2017

**Subject Name:Organic Chemistry for Technologists** 

Time: 10:30 AM to 01:00 PM

**Total Marks: 70** 

Instructions:

- 1. Attempt all questions.
- 2. Make suitable assumptions wherever necessary.
- 3. Figures to the right indicate full marks.

**MARKS** 

14

## 0.1 **Short Questions**

- 1 Define the term carbenes.
- 2 Draw the structure for 2, 2-di (p-Chlorophenyl)-1, 1, 1-trichloroethane.
- 3 Hemolytic fission of covalent bond leads to the formation of
- 4 Give the structure of Chloramine – T.
- 5 Anisole is formed when phenol is treated with .
- 6 Give the structural formula of Sulphanilamide.
- 7 Give IUPAC name of (CH<sub>3</sub>CH<sub>2</sub>)<sub>3</sub>CCN.
- 8 How many isomeric dichlorobenzene?
- 9 Hydrolysis of Benzal chloride gives\_
- CH<sub>3</sub>OH, AlCl<sub>3</sub>, NH<sub>3</sub> and CH<sub>3</sub>OCH<sub>3</sub>, which is a Lewis acid? 10
- 11 \_\_test is used to distinguish Aniline and N-methyl aniline.
- Phenol, Resorcinol and catechol; Rank these molecules in decreasing order of boiling points.
- Draw the structure of 1-ethoxy-1-propanol. 13
- 14 Define Cannizzaro reaction.
- (a) Why benzyl carboniumion is more stable than ethyl carboniumion? 0.2
- 03 (b) What is essential difference between a free radical reaction and an ionic 04
  - reaction?
  - (c) Explain Aldol and Cross aldol reaction with mechanism

**07** 

**07** 

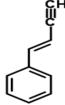
04

(c) Write a note on: Ardnt-Eister reaction with mechanism.

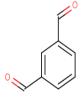
03

- Q.3 (a) Write a note on: Saccharin
  - Write the IUPAC names for each of the following compounds:
    - CH<sub>3</sub>CH<sub>2</sub>CH(OCH<sub>3</sub>)CH<sub>2</sub>COCl CH<sub>3</sub>CH<sub>2</sub>COCH<sub>2</sub>CH<sub>2</sub>COOCH<sub>3</sub> b.

c.



d.



What products are obtained by reduction of nitrobenzene under different conditions?

OR

**O.3** (a) How will you convert benzene to benzyl alcohol?

03

**07** 

	<b>(b)</b>	Write the IUPAC names for each of the following compounds:	04
		a.	
		c. OHCCH <sub>2</sub> CH <sub>2</sub> CHO d. HC=CCH <sub>2</sub> CH=CH <sub>2</sub>	
	(c)	Distinguish between SN <sup>1</sup> and SN <sup>2</sup> reaction with mechanism?	07
<b>Q.4</b>	(a)	How will you synthesis benzylamine from benzenesulfonic acid?	03
	<b>(b)</b>	Explain why,	04
		a. p-Nitrophenol is a stronger acid than phenol?	
		b. o-Nitrophenol is steam volatile whereas p-Nitrophenol is not?	
	<b>(c)</b>	Name the following reaction and Explain its mechanism & application in detail.	07
		O Br <sub>2</sub> $\begin{bmatrix} 0 & H_2O \\ 0 & H_2O \end{bmatrix}$	
		$ \begin{array}{c} O \\ R \end{array} \begin{array}{c} Br_2 \\ N = O \end{array} \begin{array}{c} O \\ -CO_2 \end{array} \begin{array}{c} R - NH_2 \end{array} $	
		OR	
<b>Q.4</b>	(a)	Two isomeric compounds (A) and (B) have molecular formula C <sub>7</sub> H <sub>8</sub> O.	03
		Compound (A) readily dissolves in aqueous NaOH, but compound (B) does not.	
		Compound (A) is insoluble in aqueous NaHCO3 and reacts with bromine water	
		form (C), C <sub>7</sub> H <sub>5</sub> OBr <sub>3</sub> . Identify (A), (B) and (C).	
	<b>(b)</b>	How will you convert Nitrobenzene into 4, 4'-Benzidine?	04
	<b>(c)</b>	How does aniline react with;	07
		a. Acetic anhydride b. Bromine water c. Benzaldehyde	
		d. Conc. $H_2SO_4$ at $180^0$ C e. Acetyl chloride f. Nitrous acid at $0^0$ to $5^0$ C.	
		g. Chloroform and alc. KOH	
<b>Q.5</b>	(a)	•	03
		operative.	
	<b>(b)</b>	·	04
		a. Aniline →m-Nitroaniline	
	( )	b. Aniline → p-Nitroaniline	0=
	(c)	Name the following reaction and Explain its mechanism & application in detail.	07
		CI NaOH	
		OR	
Q.5	(a)	Explain Pinner reaction with mechanism.	03
	<b>(b)</b>	Explain Baeyer Villiger oxidation reaction with mechanism.	04
	<b>(c)</b>	Who were the pioneers of Alkylation & Acylation reaction? Explain its	07
		mechanism.	

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