## **GUJARAT TECHNOLOGICAL UNIVERSITY** BE – SEMESTER – VI (NEW).EXAMINATION – WINTER 2016

Subject Code: 2162108 Date: 26/10/2			2016	
T	Subject Name: Material Degradation and Prevention Time: 10:30 AM to 01:00 PM Total Marks: 70 Instructions:			
		<ol> <li>Attempt all questions.</li> <li>Make suitable assumptions wherever necessary.</li> <li>Figures to the right indicate full marks.</li> </ol>		
Q.1	(a) (b)	Define Corrosion & explain electrochemical theory of wet corrosion Give the importance of corrosion study; including cost of corrosion	2+5 07	
Q.2	(a)	Derive Nernst equation for electrode potential. Show how the electrode potential of Hydrogen electrode will vary with pH at one atmospheric pressure	5+2	
	<b>(b</b> )	Write a short note on the Ag-AgCl and Calomel reference electrode. OR	7	
	<b>(b)</b>	Compare between E.M.F. & Galvanic series for corrosion studies	7	
Q.3	(a)	What is intergranular corrosion? Compare between Knife Line Attack & Weld Decay	3+4	
	<b>(b</b> )	Write Piling-Bedworth ratio and its applications OR	7	
Q.3	(a)	Explain use of material selection to combat corrosion in high temperature conditions.	7	
	<b>(b</b> )	Justify the role of design aspects in corrosion protection with suitable examples	7	
Q.4	(a)	protection, with suitable examples. Differentiate between cathodic and	4+3	
	<b>(b</b> )	anodic inhibitors During electroplating, pH of bath is strictly maintained. Give reason. <b>OR</b>	7	
Q.4	(a) (b)	What is anodizing? Give applications Explain how rusting of iron is prevented by galvanization	7 7	
Q.5	(a)	What do you mean by Tafel plot? Explain the use of Tafel plot in corrosion rate measurement	3+4	
	<b>(b)</b>	Explain how material selection is helpful to combat corrosion in Automobile Sector.	7	
_		OR	_	
Q.5	(a)	What is Wear? What are the methods to improve wear resistance? Give suitable examples.	2+5	
	<b>(b</b> )	Discuss the Wear Measurement Technique with Necessary schematics.	7	

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