GUJARAT TECHNOLOGICAL UNIVERSITY B. E. - SEMESTER – III • EXAMINATION – WINTER 2012

Subject Name: Basic Rubber Science Time: 10.30 am – 01.00 pm Total Ma	mbra. 70
-	rks: 70
 Instructions: Attempt all questions. Make suitable assumptions wherever necessary. Figures to the right indicate full marks. 	
Q.1 (a) Discuss in detail about the chain structure and chemical reactivity of	ubber. 07
 Q.1 (b) Answer the following i Write down the general rules for polymer solubility. ii Explain the term poisson's ratio. Also give the poisson's ratio rubber. 	04 alue for 03
Q.2 (a) Discuss in detail about the applications of colloids.	07
 Q. 2 (b) Answer the following i State Archimedes principle and distinguish between the terms specificand density. Describe a method for determination of relative density of soft pure gum vulcanized rubber, indicate clearly the precaution would take to minimize the error. 	of piece
ii For rubber, summarize first law of Amonton's friction. OR	02
 Q. 2 (b) Answer the following i Differentiate the rubbery deformation and elastic deformation. ii Write about the sliding friction of rubber. 	04 03
 Q.3 (a) Describe, with the aid of diagram, what you understand electromagnetic wave. Name and define the quantities which are characterize an electromagnetic wave. What are the principle reg which the electromagnetic spectrum is divided? Which of these reg found application in rubber science and technology and for what purpose. 	used to ons into ons have
 Q.3 (b) Answer the following i Write a short note on electrical properties of rubber. ii Explain the term surface tension. What will be the height of lie capillary tube of diameter 5*10⁻⁴ meter? Surface tension of water is 7*10⁻² N/meter and angle of contact is 0°. 	05 uid in a 03
Q.3 (a) Discuss the features of sinusoidal vibrations.	06
 Q. 3 (b) Answer the following i Define the following terms: (i) Conduction (ii) Convection (iii) Radii Thermal Conductivity (v) Thermal Diffusivity 	tion (iv) 05
ii A capillary tube of diameter 0.5 mm is dipped vertically in a liquid	рто ensity of 03

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		0.89 gram/cc. Determine the height of liquid in capillary. Surface tension of liquid is 28 dyne/cm and angle of contact is 16°28'.		
Q. 4	(a)	Discuss in detail about the emulsion polymerization technique.	07	
Q. 4	(b) i ii	Answer the following Differentiate the addition polymerization and condensation polymerization. Explain the term degree of polymerization with suitable example. OR	04 03	
Q. 4	(a)	Discuss the solution polymerization technique with its advantages and disadvantages.	07	
Q. 4	(b)	Answer the following		
	i	Write a short note on initiators.	05	
	ii	Give the classification of polymer based on the origin.	02	
Q. 5	(a)	List the various purification methods for colloidal solution. Describe any two.	06	
Q. 5	(b)	Answer the following		
	i	Write in detail about the characteristics of colloidal state.	04	
	ii	What do you mean by lyophilic sols? List the characteristics of it with examples.	04	
OR				
Q. 5	(a)	Write a short note on emulsion.	06	
Q. 5	(b)	Answer the following		
	i	What do you mean by tydnall effect? What are the causes of it?	04	
	ii	Write in detail about the gel. ************************************	04	