GUJARAT TECHNOLOGICAL UNIVERSITY BPHARM – SEMESTER II • EXAMINATION – SUMMER • 2014 Subject code: 220006 Date: 06-06-2014 **Subject Name: Physical Pharmacy** Time: 02:30 pm - 05:30 pm **Total Marks: 80 Instructions:** 1. Attempt any five questions. 2. Make suitable assumptions wherever necessary. 3. Figures to the right indicate full marks. Classify complexes and write on metal complexes in detail. Q.1 06 **(a)** Differentiate between channel lattice type and clathrates complexes. **(b)** 05 Enumerate methods of analysis for complexes. Write on pH titration method. (c) 05 Q.2 **(a)** Write in brief on any two: **06** i) Protein binding ii) Pharmaceutical application of colloids iii) Association colloids What are kinetic properties of colloids? 05 **(b)** Explain electrical properties of colloids. (c) 05 **Q.3** Write on: **(a)** 06 i)Particle and particle size distribution ii)Number and weight distribution Explain derived properties of powders. 05 **(b)** Give in brief methods to determine particle shape and surface area. **(c)** 05 **O.4** Give Newton's law of flow. Define kinematic viscosity. Give relation between 06 **(a)** temperature effect on viscosity of liquids. **(b)** Write briefly on any one: i) Plastic Flow ii) Pseudoplastic Flow iii) Dilatant Flow 05 Write on Thixotropy, measurement of thixotropy and its applications in 05 (c) pharmaceutical formulations. Q.5 Write in short on any one: 06 (a) i) Sedimentation of flocculated particles ii) Difference between flocculated and deflocculated suspensions. iii) Formulation of suspensions. Give types, applications and preservation of emulsions. **(b)** 05 Write on Rheological properties of emulsions. (c) 05 Write in brief on (any two): Q. 6 **(a) 06** 1. Spreading Coefficient 2. CMC 3. Surface Active Agents 4. Systems of HLB classification Write on: Surface and interfacial tensions and techniques of their measurement. Write 05 **(b)** in brief on Surface Free Energy. Write in details on: Physical Stability of emulsions. (c) 05 Q.7 Explain Liquefaction of gases and Methods of achieving liquefaction. Give **(a)** 06 application of this phenomenon in Pharmacy. Explain the following(any one): 05 **(b)** i) Crystalline solids ii) Polymorphisms iii) Amorphous solids Discuss 2 component systems containing liquid phases or Ternery systems with one (c) 05 pair of partially miscible liquids.
