## GUJARAT TECHNOLOGICAL UNIVERSITY B. PHARM. - SEMESTER - VII • EXAMINATION - WINTER 2012

Subject code: 270004 Date: 03/01/2013			
Subject Name: Pharmaceutical Analysis - III Time: 10.30 am - 01.30 pm Total Marks: 80 Instructions:			
Ins	1. 2.	Attempt any five questions.  Make suitable assumptions wherever necessary.  Figures to the right indicate full marks.	
Q.1	(a) (b)	State and explain Beer's law. Discuss the factors leading to deviation from this law. Give an account of the detectors used in UV VIS spectrophotometer. Add a note on monochromators.	06 05
	(c)	10.00 mg of a drug sample was dissolved in water and volume was adjusted to 1 Ltr. with the same solvent. The absorbance was found to be 0.556 at its $\lambda_{max}$ . Calculate purity of the sample (M. Wt. 200.0 and molar absorptivity is 1.32 X 10 <sup>4</sup> ).	05
Q.2	(a)	Explain various transitions occurring in a molecule when electromagnetic radiations interact with it. Discuss in detail the principle of IR Spectroscopy.	06
Q.3	(b) (c) (a)	Discuss constructions and working of Michelson interferometer.  Give a detailed account of various regions of electromagnetic spectrum.  Explain the theory of fluorescence and phosphorescence. Discuss the factors affecting fluorescence intensity.	05 05 06
Q.4	(b) (c) (a) (b) (c)	Draw a well labeled diagram of Spectrofluorimeter. Explain advantages and limitations of fluorescence spectroscopy.  Discuss sample handling in UV – VIS and IR Spectroscopy.  Explain the principle of NMR. Give an account of instrumentation in NMR.  Define chemical shift. Explain in brief factors affecting chemical shift.  Write a short notes: (Any TWO)  1. Spin-Spin coupling 2. Coupling constant 3. TMS as an internal standard	05 06 05 05
Q.5	(a) (b)	Discuss the theory of Mass Spectroscopy. Give an account of Ionization techniques used in Mass spectroscope.  Draw a well labeled diagram of a Mass Spectroscope. Add a note on fragmentation	06
	(c)	rules. Enlist different Mass analyzers and discuss any one in detail.	05
Q. 6	(a) (b) (c)	Explain the principle of Atomic absorption Spectroscopy. Give its applications. Give a detailed account of interferences in Atomic Spectroscopy.  Write a short notes: (1) Hollow Cathod lamp (2) Flame photo meter	06 05 05
Q.7	(a) (b) (c)	Discuss simultaneous equation method for analysis of Binary Mixture.  Describe the factors influencing vibrational frequencies.  Write Short note (ANY TWO)  1. C13 NMR	06 05 05