GUJARAT TECHNOLOGICAL UNIVERSITY M. E. - SEMESTER – II • EXAMINATION – SUMMER • 2014

Subject code: 1723001

Date: 16-06-2014

Subject Name: Advance Mass Transfer

Time: 02:30 pm - 05:00 pm

Instructions:

Total Marks: 70

- 1. Attempt all questions.
- 2. Make suitable assumptions wherever necessary.
- 3. Figures to the right indicate full marks.

Q.1

A multicomponent distillation column should be designed in such a 14 way, that a concentrate stream of Ethane could be separated from a feed stream containing 6 hydrocarbons. Calculate the required theoretical number of stages by FUG method, when R = 1.25Rmin. Separation requirements:

Mole fraction of Propylene in the Distillate: 0.025 Mole fraction of Ethane in the Bottoms: 0.050

Operating conditions:

Pressure: p = 27.3 bar

Feed quality: liquid at the boiling temperature (q = 1) Average plate efficiency coefficient: E = 0.75

Feed composition:

Components in the	mol-%	Feed
Feed stream		
Methane	5.0	7.356
Ethane (LK)	35.0	2.091
Propylene (HK)	15.0	1.000
Propane	20.0	0.901
i-Butane	10.0	0.507
n-Butane	15.0	0.408

Mole flow-rate for a Feed of 100 kmol/h: Assumption: The non-key components are completely separated.

- Q.2 (a) Discuss the Process design steps of Falling film absorber. 07
 - (b) Write a Shot note on Thermally coupled Distillation Column. 07

OR

- (b) Discuss the use of heat pump with refrigerant in distillation column 07 for energy saving.
- Q.3 By using Lewis Matheson method, calculate the composition of 14 vapour entering top most tray of the column for following case.(a) Distillation columns details:

Component	Feed,	Distillate	Residue
	Mole%	mole %	mole %
n-Butane	37	95	16.3
i-Pentane	32	05	41.6
n-Pentane	21		28.5
n-Hexane	12		13.6

- (b) F = 4750 kmol/hr, D = 1250 and feed is saturated liquid
- (c) Operation Pressure = 2 atm ab

- (d) Reflux ration R = 3
- (e) Vapour pressure of components:

Component	А	В	С
n-butane	15.6782	2154.9	-34.42
i- pentane	15.6338	2348.67	-40.05
n-pentane	15.8333	2477.07	-39.94
n-hexane	15.8366	2697.55	-48.78

 $\ln Pv = A \circ (B/T+C)$, where Pv is in mmHg & Temp. in K

OR

- Q.3 Explain the stepwise procedure multicomponent batch distillation 14 with constant overhead composition.
- Q.4 Predict the Fractional solute removal and pressure drop in a venturi 14 scrubber based on the following data.
 - (1) Volumetric flow rate of boiler flue gas = $24000 \text{ Nm}^3/\text{h}$
 - (2) Discharge pressure of gas from venture = Atmospheric
 - (3) Temperature of gas = $90^{\circ}C$
 - (4) SO₂ concentration in boiler flue gas = 4000 ppm
 - (5) Solvent = 1% lime solution
 - (6) Solvent to gas ratio = 1.4 L/m^3
 - (7) Throat velocity of gas phase = 100 m/s
 - (8) Average molar mass of flue gas = 29.48 kg/kmol
 - (9) Equilibrium mass of SO₂ per 100 mass of $H_2O = 0.035$

Assume that no Chemical reaction is taking place between lime and SO_2 due to presence of 1% lime in solution. Also assume the average temperature of scrubber is 50°C.

Reliable pressure drop equation for venturi scrubber is,

 $P = 2.584*10^{-3} V_{G} \quad _{G} A_{th} (L \emptyset / G \emptyset)^{0.78}$

OR

- Q.4 (a) Explain the method of selection of operating pressure in distillation 07 column. Also define light key and heavy key components.
 - (b) Discuss the criteria for selection of Spray tower, Venturi scrubber 07 and Tray tower as absorbers.
- Q.5 (a) Describe the advantages and disadvantages of microfiltration.
 (b) Discuss the Applications of Super heated Steam dryer in detail.
 OR
- Q.5 Write advantages and disadvantages of superheated steam dryer 14 over hot air dryer.
