Enrolment No._____

GUJARAT TECHNOLOGICAL UNIVERSITY M. Pharm. – SEMESTER – II • EXAMINATION – SUMMER • 2014

$\mathbf{M}, \mathbf{I} \text{ harm}, = \mathbf{SEWESTEK} = \mathbf{H} \cdot \mathbf{EXAMINATION} = \mathbf{SOWMEK} \cdot 2014$			
Subject Code: 2920101 Date: 29-05-2014			
Subject Name: Advanced Organic Chemistry - II Time: 02:30 pm - 05:30 pm Total Marks: 80			
Instructions:			
		Attempt any five questions.	
		Make suitable assumptions wherever necessary. Figures to the right indicate full marks.	
01			08
Q.1	(a)	Explain in detail about the principle and reaction mechanism of Bayer-villiger rearrangement.	Uð
	(b)	What is the importance of stereochemistry in Drug Discovery? Explain in context to geometrical and optical isomerism with suitable examples.	08
Q.2	(a)	Give the rules and guidelines used in reterosynthesis	04
	(b)	Give synthon approach for following drugs.	12
		a) Diclofenacb) Cetrizine	
		c) Losartan	
Q.3	(a)	How stereochemistry can be useful for improvement of pharmacokinetic and	04
		pharmacodynamic of the drugs	10
	(b)	Give asymmetric synthesis of a) Ampicillin	12
		b) Omeprazole	
		c) Nifedipine	
		d) Ethambutol	
Q.4	(a)	Write a note on the following a) Swern oxidation	4x4
		b) Reformastsky reaction	
		c) Curtius rearrangement	
		d) Heck reaction	
Q.5	(a)		08
	(b)	Write a note on ionic liquids and supercritical fluid.	08
Q. 6	(a)	What is difference between conventional and microwave heating? Explain in detail about principle and solvent selection criteria in microwave.	08
	(b)	Write a detailed note on stereochemistry of compound having asymmetric	08
		planes.	
Q.7	(a)	Write a note on the following;	4x4
		a) Solvent free reactionb) Nanochemistry	
		c) Conformational isomers of dimethyl cyclohexane	
		d) Racemic switches	
