## **GUJARAT TECHNOLOGICAL UNIVERSITY** PDDC - SEMESTER-I • EXAMINATION – SUMMER 2013

Subject Code: X11902Date: 11-06-2013Subject Name: Engineering Thermodynamics			;
Time: 02.30 pm - 05.00 pm Total Marks: 7 Instructions:			)
instru	1. A 2. N	ttempt all questions. Iake suitable assumptions wherever necessary. igures to the right indicate full marks.	
Q.1		<ul> <li>What do you mean by thermodynamics? Explain the concepts of microscopic and macroscopic point of view of thermodynamics.</li> <li>Define the following terms:- <ol> <li>Process (2) Cycle (3) Pure substance</li> <li>Control volume (5) Closed System</li> <li>Intensive Properties (7) Extensive Properties</li> </ol> </li> </ul>	07 07
Q.2	(a)	Explain Joule's Experiment for first law of thermodynamics with neat sketch.	07
	<b>(b)</b>	Derive and explain steady flow energy equation. OR	07
	<b>(b</b> )	Explain the concept of perpetual motion machine of the first kind.	07
Q.3	(a)	What are the various limitations of first law of thermodynamics? Explain in detail with various examples.	07
	(b)	An Engine operating on Carnot cycle works within temperature limits of 700 K and 350 K. If the engine receives 1800 kJ of heat , Calculate the following:- (1) Thermal efficiency (2) Work done (3) Heat Rejected.	07
		OR	
Q.3	(a)	In a gas turbine unit, the gas flow through the turbine is 14 kg/s and the power developed by the turbine is 11500 kW. The enthalpies of the gases at the inlet and outlet are 1250 KJ/kg and 380 kJ/kg respectively, and the velocity of gases at the inlet and outlet are 45m/s and 100m/s respectively. Calculate the following:- (1) The rate at which heat is rejected to the turbine	07
	(b)	(2) The area of inlet pipe given that the specific volume of the gases at the inlet is $0.40 \text{ m}^3/\text{kg}$ .	07
Q.4	(a)	What do you mean by Availability? Derive an expression for availability	07
	<b>(b</b> )	in non-flow systems. Determine the air standard efficiency of Otto cycle from the following data	07

data. Bore of the cylinder = 13 cm Stroke Length = 12 cm Clearance volume =  $300 \text{ cm}^3$ , Take  $\gamma$ =1.4

- Q.4 (a) Compare the Carnot Cycle with Rankine Cycle.
- (b) A vessel of volume  $0.38 \text{ m}^3$  consists of 0.40 kg of CO and 1 kg of Air at 07 Q.4 14 °C.Calculate (1) The partial pressure of each constituent (2) Total pressure in the vessel. The air contains 23 % O<sub>2</sub> and 76.9 % N<sub>2</sub> by mass. Take the molecular mass of CO, O2, N2 as 28, 32 and 28 kg/kg mole respectively. Take Universal Gas constant, R=8.3143 kJ/kg mol k Q.5 What do you mean by Calorific value? Explain Junker's Gas Calorimeter 07 (a) with neat sketch. 07 (b) State and explain the following:-(1) Avogadro's law (2) Gibbs-Dalton Law OR (a) Explain Diesel Cycle with neat sketch and derive the equation for the Q.5 07 efficiency of Diesel cycle. (b) State and explain the following in brief:-07 (1) Third Law of thermodynamics (2) Entropy and disorder

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